

**AMENDMENTS TO THE CLAIMS**

Claims 1 - 10 (previously cancelled)

Claim 11 (previously amended) A process for preparing a defective metal oxide for a battery cathode with increased lithium capacity, said process comprising:

providing a sufficient amount of metal oxide;

heating said metal oxide under an atmosphere consisting essentially of  $O_2$  and  $H_2O_{(g)}$  gas;

and

cooling said metal oxide, wherein said heating under said atmosphere introduces local ionic defects and increases the lithium capacity of said metal oxide.

Claims 12-16 (previously canceled)

Claim 17 (previously amended): The process as in claim 11, further comprising the step of:

maintaining said heating at a temperature of from about 300 to about 600 °C.

Claim 18 (previously amended): The process as in claim 17, wherein said heating is maintained from about 6 to about 72 hours.

Claim 19 (previously amended): The process as in claim 11, wherein said  $O_2$  and  $H_2O_{(g)}$  is applied to said metal oxide sample at a linear flow rate of about 50 ccm to about 350 ccm.

Claim 20 (previously amended): The process as in claim 11, wherein said heating is from about 2 to about 20 °C/min up to about 460 °C.

Claim 21 (previously presented): The process as in claim 20, further comprising the step of:  
maintaining said temperature of about 460 °C for 24 hours.

Claim 22 (previously presented): The process as in claim 11, wherein said cooling is from about 2 to about 20 °C/min until ambient air temperature is achieved.

Claim 23 (previously amended): The process as in claim 11, wherein said metal oxide is  $V_2O_5$ .

Claim 24 (previously amended): The process as in claim 11, wherein said metal oxide comprises a surface area of about 1- 10 square meters.

Claim 25 (previously amended) A process for preparing a defective metal oxide for a battery cathode with increased lithium capacity, said process comprising:

providing a sufficient amount of metal oxide;

heating said metal oxide at a temperature of from about 300 to about 600 °C for a time period of from about 6 to about 72 hours under an atmosphere consisting essentially of  $O_2$  and  $H_2O_{(g)}$  gas, wherein said  $O_2$  and  $H_2O_{(g)}$  is applied to said metal oxide sample at a linear flow rate of about 50 ccm to about 350 ccm; and

cooling said metal oxide, wherein said heating under said atmosphere introduces local ionic defects and increases the lithium capacity of said metal oxide.